Study program: Integrated academic studies of Pharmacy					
Type and level of the study program: integrated academic studies					
Course title: Preparative analytical chemistry in pharmacy (PhIII-PACH)					
Teacher: Nataša B. Milić, Nataša P. Milošević, Maja Lj. Milanović, Mira P. Mikulić					
Course status: elective					
ECTS Credits: 3					
Condition: Organic Chemistry 2; Analytical Chemistry 2; Instrumental Pharmaceutical Analysis					
Course aim					
The aim of this course is to introduce students with preparatory analytical methods and give them the knowledge necessary to select the best method					
to prepare samples for analysis.					
Expected outcome of the course:					
Mastering the theoretical aspects of analytical methods for sample preparation, introduction to the basic principles of instruments used in preparative					
chemistry analysis and their advantages and disadvantages.					
Knowledge of various preparative and analytical methods for selection of the most appropriate methods for sample preparation.					
Course description			0. Membrane extraction.		
Theoretical education		21.	1. Preparation of samples for analysis of metals.		
1. Errors in the quantitative analysis. The accuracy and precision.			. Wet digestion.		
2. Sample preparation. Sample storage.			. Dry ashing method.		
3. Quality control in the process of sample preparation.			A. Preparation of water samples.		
4. Extraction and preconcentration of the diluted sample.			5. Methods of precipitation.		
5. Principles of extraction and the extraction of semi-volatile organic			Preparation of sediment for direct atomic absorption spectroscopy.		
compounds from liquid samples.			. Colorimetric methods.		
6. Liquid-liquid extraction. Liquid-solid extraction.		28.	8. Contamination of the sample during the analysis of metals.		
7. Solid phase extraction-SPE.					
8. Solid phase microextraction.		Practical education: exercises, other forms of education, research			
9. Stir bar sorptive extraction - SBSE.		related activities			
10. Principles of extraction and the extraction of semi-volatile organic		Selected examples of theoretical and experimental exercises:			
compounds from solid samples.		1. Theoretical comparison of extraction methods of semi-volatile			
11. Soxhlet extraction.		2.	organic compounds from liquid samples.		
<ol> <li>12. Ultrasonic extraction.</li> <li>13. Supercritical fluid extraction.</li> </ol>		۷.	. Theoretical comparison of extraction methods of semi-volatile organic compounds from solid samples.		
14. Microwave extraction.		3.			
15. Extraction with high pressure and temperature (Accelerated Solvent		5.	organic compounds from solid and liquid samples.		
Extraction -ASE)		4.			
16. Extraction of volatile organic compounds from solid and liquid		т.	analysis of metals.		
samples.		5.			
17. Static headspace extraction.		volatile organic compounds from liquid samples.			
18. Dynamic headspace extraction of purge and trap.		6. Selected experimental exercises of the extraction method for semi-			
19. Liquid-liquid extraction of large volume.		volatile organic compounds from solid samples.			
		7. Selected experimental exercises of the extraction method for			
		volatile organic compounds from solid and liquid samples.			
		8.			
			for the analysis of metals.		
Literature					
Compulsory					
1. Somenath M. Sample preparation techniques in analytical chemistry. John Wiley & sons, Inc, Publication Hoboken, New Jersey, 2003,					
Additional					
1. Internal script for practical education.					
Number of active classes					Other:
	er types of teaching:		Research related activities:		
30 15					
Teaching methods: lectures, interactive classes, e					
	tudent activity assessment	t (m			
Pre-exam activities points			Final exam		points
Lectures	5		Written		50
Practices	20		Oral		
Colloquium					
Essay	25				